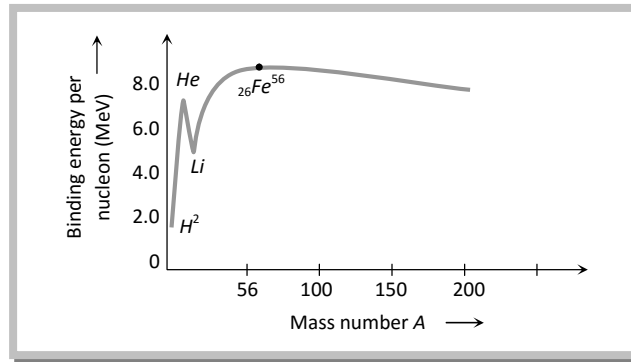


## Binding Energy Curve.

It is the graph between binding energy per nucleon and total number of nucleons (i.e. mass number A)



(1) Some nuclei with mass number  $A < 20$  have large binding energy per nucleon than their neighbor nuclei. For example  ${}^2\text{He}^4$ ,  ${}^4\text{Be}^8$ ,  ${}^6\text{C}^{12}$ ,  ${}^8\text{O}^{16}$  and  ${}^{10}\text{Ne}^{20}$ . These nuclei are more stable than their neighbors.

(2) The binding energy per nucleon is maximum for nuclei of mass number  $A = 56$  ( ${}_{26}\text{Fe}^{56}$ ). Its value is 8.8 MeV per nucleon.

(3) For nuclei having  $A > 56$ , binding energy per nucleon gradually decreases for uranium ( $A = 238$ ), the value of binding energy per nucleon drops to 7.5 MeV.

Note: When a heavy nucleus splits up into lighter nuclei, then binding energy per nucleon of lighter nuclei is more than that of the original heavy nucleus. Thus a large amount of energy is liberated in this process (nuclear fission).

When two very light nuclei combines to form a relatively heavy nucleus, then binding energy per nucleon increases. Thus, energy is released in this process (nuclear fusion).

