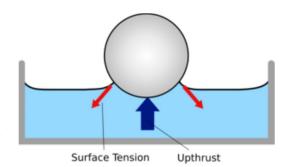
Properties Of Solid And Liquid

Liquids at Rest:-

- Force of cohesion:- It is force between two molecules of similar nature.
- Force of adhesion:- It is the force between two molecules of different nature.
- **Molecular range:** The maximum distance between two molecules so that the force of attraction between them remains effective is called molecular range.
- Sphere of influence:- Sphere of influence of any molecule is the sphere with molecule as its center and having a radius equal to molecular range (= 10^{-7} cm).
- **Surface film:** Surface film of a liquid is defined as the portion of liquid lying on the surface and caught between two parallel planes situated molecular range apart.
- Surface tension:-

Surface tension is the property of a liquid by virtue of which its free surface behaves like a stretched membrane and supports, comparatively heavier objects placed over it. It is measured in terms of force of surface tension.



• Force of surface tension:- It is defined as the amount of force acting per unit length on either side of an imaginary line drawn over the

liquid surface.

- (a) T = Force/length = F/l
- (b) T = Surface energy/Surface area = W/A

Units:- S.I - Nm⁻¹

C.G.S- dyn cm⁻¹

• Additional force:-

- (a) For a cylindrical rod:- $F = T \times 2\pi r$ (Here r is the radius of cylindrical rod)
- (b) For a rectangular block:- $F = T \times 2(l+d)$ (Here l is the length and d is the thickness of the rectangular block)
- (c) For a ring:- $F = T \times 2 \times 2\pi r$ (Here r is the radius of cylindrical rod)
 - Surface energy:-

Potential energy per unit area of the surface is called surface energy.

(a) Expansion under isothermal condition:-

To do work against forces of surface tension:-

 $W = T \times A$ (Here A is the total increase in surface area)

To supply energy for maintaining the temperature of the film:-

E = T + H

(b) Expansion under adiabatic conditions:-

E = T

Force of surface tension is numerically equal to the surface energy under adiabatic conditions.

- Drops and Bubbles:-
- (a) Drop:- Area of surface film of a spherical drop of radius *R* is given by, $A = 4\pi R^2$
- (b) Bubble:- The surface area of the surface films of a bubble of radius *R* is, $A = 2 \times 4\pi R^2$
 - Combination of *n* drops into one big drop:-
- (a) $R = n^{1/3}r$
- (b) $E_{\rm i} = n \ (4\pi r^2 T)$, $E_{\rm f} = 4\pi R^2 T$
- (c) $E_{\rm f}/E_{\rm i}=n^{-1/3}$
- (d) $\Delta E/E_i = [1-(1/n^{1/3})]$
- (e) $\Delta E = 4\pi R^2 T (n^{1/3} 1) = 4\pi R^3 T (1/r 1/R)$
 - Angle of contact:- Angle of contact, for a pair of solid and liquid, is defined as the angle between tangent to the liquid surface drawn at the point of contact and the solid surface inside the liquid.
- (a) When θ < 90° (acute):-

$$F_a > F_c / \sqrt{2}$$

- (i) Force of cohesion between two molecules of liquid is less than the force of adhesion between molecules of solid and liquid.
- (ii) Liquid molecules will stick with the solid, thus making solid wet.
- (iii) Such liquid is put in the solid tube; it will have meniscus concave

upwards.

- (b) When $\theta > 90^{\circ}$ (obtuse):- $F_a < F_c / \sqrt{2}$
- (i) Force of cohesion between two molecules of liquid is less than the force of adhesion between molecules of solid and liquid.
- (ii) In this case, liquids do not wet the solids.
- (iii) Such liquids when put in the solid tube will have a meniscus convex upwards.
- (c) When $\theta = 90^{\circ}$:-?

$$F_a = F_c / \sqrt{2}$$

The surface of liquid at the point of contact is plane. In this case force of cohesion and adhesion are comparable to each other.

(d)
$$\cos\theta_{\rm c} = T_{\rm sa} - T_{\rm sl}/T_{\rm la}$$

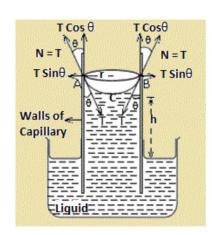
Here, $T_{\rm sa}$, $T_{\rm sl}$ and $T_{\rm la}$ represent solid-air, solid-liquid and liquid-air surface tension respectively). Here $\theta_{\rm c}$ is acute if $T_{\rm sl} < T_{\rm sa}$ while $\theta_{\rm c}$ is obtuse if $T_{\rm sl} > T_{\rm sa}$.

Capillarity:-

??Capillarity is the phenomenon, by virtue of which the level of liquid in a capillary tube is different from that outside it, is called capillarity.

Weight of liquid,
$$W = V\rho g = \pi r^2 [h + (r/3)]\rho g$$
 (Here *r* is the radius meniscus)

If weight of meniscus is taken into account,



the force of surface tension will be,

$$T = [r(h+(r/3)) \rho g]/2 \cos\theta$$

For fine capillary, force of surface tension, $T = rh\rho g/2 \cos\theta$

So height,
$$h = 2T \cos\theta / r\rho g$$

This signifies, height of liquid risen (or depressed) in a capillary tube varies inversely as the radius of tube. Smaller the diameter of capillary tube, greater is the rise of liquid in it.

• Tube of insufficient length:-

$$Rh = 2T/\rho g$$

As, T, ρ and g are all constant, Rh = Constant

Smaller the value of *h*, greater will be the value of *R*. But liquid will never flow.

• Effect of temperature affecting surface tension of liquids:-

Surface tension of a liquid decreases with an increase in its temperature.

$$T_{\theta} = K (\theta_c - \theta)$$

Here T_{θ} is the surface tension at a particular temperature θ while θ_c is the critical temperature of the liquid and K is constant.

• Effect of density:- Density of liquid also affects its surface tension. Surface tension of a liquid is given by,

$$T = A (\rho - \rho')^n$$

Here, ρ is the density of liquid, ρ' is the density of saturated vapors of liquid and A is the constant depending on the nature of liquid.

• Pressure difference across a liquid surface:-

(a) Plane surface:- There is no difference of pressure on the two sides of the film.

(b) Convex surface:-Pressure below the surface film must be greater than that just above it.

(c) Concave surface:- Pressure on the upper side is greater than that just below it.

• General formula for excess pressure:-

$$P_{\text{excess}} = T[1/R_1 + 1/R_2]$$

• Excess pressure in liquid drop:-

 $P_{\text{excess}} = 2T/R$, Here *R* is the radius of liquid drop.

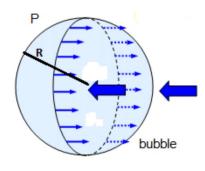
• Excess pressure for an air bubble in liquid drop:-

$$P_{\text{excess}} = 2T/R$$

• Excess pressure in soap bubble:-

 $P_{\text{excess}} = 4T/R$, Here *R* is the radius of soap bubble.

 Pressure inside an air bubble at a depth h in a liquid:- P_{in} = P_{atm}+ hdg + (2T/R)



• Forces between two plates with thin water film separating them:-

(a)
$$\Delta P = T (1/r - 1/R)$$

(b)
$$F = AT (1/r - 1/R)$$

- (c) If separation between plate is d, then $\Delta P = 2T/d$ and F = 2AT/d
 - Radius of curvature of common film:- $R_{\text{comon}} = rR/R$ -r
 - Capillary depression, $h = 2T \cos (\pi \theta) / rdg$
 - Shape of liquid surface:-
- (a) Plane surface (as for water silver) if $F_{\text{adhesive}} > F_{\text{cohesive}} / \sqrt{2}$
- (b) Concave surface (as for water glass) if $F_{\text{adhesive}} > F_{\text{cohesive}} / \sqrt{2}$
- (c) Convex surface (as for mercury-glass) if $F_{\rm adhesive} < F_{\rm cohesive} / \sqrt{2}$
 - Increase in temperature:-

$$\Delta\theta = 3T/\rho s (1/r - 1/R)$$
 or $\Delta\theta = 3T/\rho s J (1/r - 1/R)$

Flow of Liquids and Viscosity

(Mechanical Properties of fluids):-

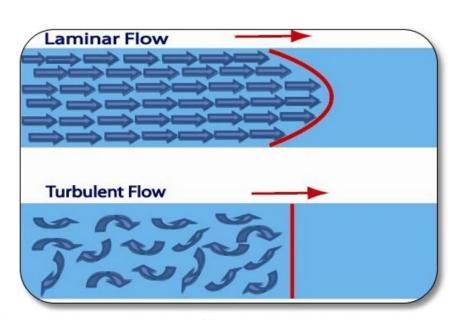
Characteristic of Ideal fluid:-

- (a) It is incompressible
- (b) It is non-viscous
- (c) Flow of ideal fluid is irrational
- (d) It is capable of exhibiting steady flow

Stream line flow:- Flow of a liquid fluid is said to be streamlined if the velocity of a molecule, at any point, coincides with that of the preceding one.

Tube

• of flow:- A bundle of



streamlines having same velocity of fluid elements, over any crosssection perpendicular to the direction of flow, is called a tube of flow.

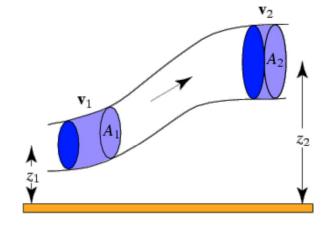
- Laminar flow:- It is a special case of streamline flow in which velocities of all the molecules on one streamline is same throughout its motion.
- **Turbulent flow:-** Whenever the velocity of a fluid is very high or it rushes past an obstacle so that there is a sudden change in its direction of motion, the motion of fluid becomes irregular, forming eddies or whirlpools. This type of motion of fluid is called turbulent flow.
- Rate of flow (Equation of continuity):-

$$a_1v_1 = Constant (a_1v_1 = a_2v_2)$$

Equation of continuity can be considered to be a statement of conservation of mass.

So,
$$v \propto 1/a$$

Velocity of flow of liquid varies inversely as the area of cross-



section of the opening from where the liquid comes out.

- Total energy of a liquid:-
- (a) Kinetic energy:- It is the energy possessed by a liquid by virtue of its velocity.

$$K.E = \frac{1}{2} mv^2$$

K.E per unit mass =
$$\frac{1}{2}$$
 v²

K.E per unit volume =
$$\frac{1}{2}$$
 [mv²/V] = $\frac{1}{2}$ ρ v²

Here, ρ is the density of liquid.

(b) Potential energy:- It is the energy possessed by a liquid by virtue of which of its position.

Potential energy = mgh

P.E per unit mass = mgh/m = gh

P.E per unit volume = $mgh/V = \rho gh$

(c) **Pressure energy:-** It is the energy possessed by a liquid by virtue of its pressure.

Pressure energy = $p \times V = m (p/\rho)$

Pressure energy per unit mass = p/ρ

Pressure energy per unit volume = $p \times V /V = p$

• **Total energy:-** Total energy of a liquid is the sum total of kinetic energy, potential energy and pressure energy.

$$E = \frac{1}{2} mv^2 + mgh + mp/\rho$$

Total energy per unit mass = $\frac{1}{2}v^2 + gh + p/\rho$

Total energy per unit volume = $\frac{1}{2} \rho v^2 + \rho gh + p$

- **Bernoulli's equation:-** It states that the total energy of a small amount of an incompressible non-viscous liquid flowing without friction from one point to another, in a streamlined flow, remains constant throughout the displacement.
- (a) $\frac{1}{2}$ mv² + mgh+ mp/ ρ = Constant
- (b) $\frac{1}{2} v^2 + gh + p/\rho = Constant$
- (c) $\frac{1}{2} \rho v^2 + \rho g h + p = Constant$ or $v^2/2g + h + p/\rho g = Constant$

The term $v^2/2g$ is called velocity head, h is called gravitational head and p/pg is called pressure head.

Therefore Bernoulli's theorem states that in case of an incompressible, non-viscous fluid, flowing from one point to another in a streamlined flow, the sum total of velocity head, gravitational head and the pressure head is a constant quantity.

Limitation of Bernoulli's equation:-

- (a) Force of viscosity, which comes into play in case of fluids in motion has not been accounted for.
- (b) Loss of energy due to heat is not accounted for.
- (c) When a fluid flows in a curved path, the energy due to centripetal force is also not accounted for.
 - If v is the relative velocity of top layer w.r.t. any other deeper layer (may be the lowest), then v is lesser for greater depth.

v = K/bd

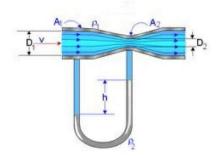
or $v \propto 1/d$

• Venturimeter:-

?It is a device used for measuring the rate of flow of liquids, generally water, through pipes.

The rate of flow of water, Q = $a_1a_2\sqrt{2hg/[a_1^2-a_2^2]}$

 Torricelli's theorem (velocity of efflux):-



It states that the velocity of efflux of a liquid (V), from an orifice, is equal to the velocity acquired by a body, falling freely (v), from the surface of liquid to the orifice.

So,
$$V = v = \sqrt{2gh}$$

- **Viscosity:-** Viscosity is the property of fluids by virtue of which they tend to destroy any relative motion between their layers.
- **Velocity gradient:-** Velocity gradient is defined as the rate of change of velocity with respect to distance.
- (a) Velocity gradient = dv/dr
- (b) Dimension of velocity gradient = $[dv/dr] = [T^{-1}]$
- (c) Direction of velocity gradient is perpendicular to the direction of flow, directed in the direction of increasing velocity.
- (d) Average velocity gradient:- Average velocity gradient is the difference between velocities of two layers separated a unit distance apart.

Average velocity gradient = $\Delta v/\Delta r$

• Newton's law of viscosity:-

In accordance to Newton's law of viscosity, the viscous drag force depends upon the nature of fluid along with following factors:-

- (a) F∝A (common area of two layers)
- (b) F∝dv/dr (velocity gradient)
- (c) So, $F = \eta A (dv/dr)$

Here η is called coefficient of viscosity of fluid.

• Coefficient of viscosity of fluid (η_v) or fugitive elasticity:-

 $\eta_{\rm v}$ = shear stress/velocity gradient = (F/A)/(dv/dr)

• Modulus of rigidity(η_r):-

$$\eta_r$$
 = shear stress/shear strain = $(F/A)/(\theta)$ = $(F/A)/(dx/dr)$

Here, $\theta = dx/dr = displacement gradient$

• Coefficient of viscosity (Absolute viscosity or Dynamic viscosity):-

$$F = \eta A (dv/dr)$$
 if $A = 1$, $dv = 1$, $dr = 1$, $F = \eta$

Co-efficient of viscosity of a fluid is defined as the tangential force per unit area which is required to maintain (or resist) a unit relative velocity between two layers a unit distance apart.

Or

Co-efficient of viscosity of a fluid is defined as the tangential force per unit area which is required to maintain a unit velocity gradient between its layers.

Unit of η:-

S.I:-
$$\eta = 1$$
 deca poise = 1 N sec/m²

Co-efficient of viscosity of a fluid is said to be one deca-poise if a tangential force of 1 N per meter square is required to maintain a relative velocity of 1 ms⁻¹ between its layer 1 m apart.

C.G.S:-
$$\eta = 1$$
 poise = 1 dyn sec/cm²

Coefficient of viscosity of a fluid is said to be one poise if a tangential force of 1 dyn per square cm is required to maintain a relative velocity of 1

cms⁻¹ between its layers 1 cm apart.

• Relation between deca-poise and poise:-

• Dimension formula for η:-

$$\eta = Fdr/Adv = [M^1L^{-1}T^{-1}]$$

• **Fluidity:-** Reciprocal of coefficient of viscosity of a fluid is called its fluidity.

Fluidity =
$$1/\eta$$

Unit of fluidity: poise⁻¹

Dimension of fluidity: [M⁻¹L¹T¹]

• **Kinematic viscosity:-** Kinematic viscosity of a fluid is defined as the ration between its coefficient of viscosity to the density of fluid.

Kinematic viscosity = η/ρ

Units of kinematic viscosity:- C.G.S – 1 stoke = $cm^2 s^{-1}$

Kinetic viscosity of a fluid having its dynamic viscosity one poise and density one g cm⁻³ is said to be 1 stoke.

Dimensional formula of kinematic viscosity = $\eta/\rho = [M^0L^2T^{-1}]$

• Critical velocity (Reynold's Number):- Critical velocity (v_c) is the maximum velocity of the flow of liquid flowing in a streamlined flow.

$$v_c = N_R \eta/\rho D$$

Here η is the coefficient of viscosity of liquid, ρ is the density of liquid and

D is the diameter of the tube.

Reynold's Number, $N_R = \rho v_c D / \eta$

- Stokes law:- In accordance to Stoke's law, force of viscosity F depend upon,
- (a) Co-efficient of viscosity of fluid η
- (b) Radius of the moving body r
- (c) Velocity of body v

So, force of viscosity, $F = 6\pi \eta r v$

- Terminal velocity:- $v = 2/9 [r^2 (\rho \sigma)/\eta]$
- $\eta = 2/9 [r^2 (\rho \sigma)g/v]$
- Variation of viscosity with a change in temperature and pressure:-
- (a) Effect of temperature:-

$$\eta = A/(1+Bt)^c$$

Here A, B and C are constants.

Again,
$$\eta v^{1/2} = Ae^{c/vt}$$

Here, A and C are constants and v is the relative velocity.

- **(b) Effect of pressure:-** Co-efficient of viscosity of liquids increases due to an increase in pressure but there is no relation, so far, to explain the effect.
 - Change in viscosity of gases:-

(a) Effect of temperature:- Co-efficient of viscosity of a gas at a given temperature is given by,

$$\eta = \eta_0 A T^{1/2}$$

Here T is the absolute temperature of gas.

Modified formula,
$$\eta = [\eta_0 A T^{1/2}]/[1 + (S/T)]$$

- **(b) Effect of pressure:-** At low pressure, co-efficient of viscosity of a gas varies directly with pressure.
 - Rate of flow of liquid through a liquid through a capillary tube of radius r and length l,

$$V = \pi pr^4/8\eta l = p/(8\eta l/\pi r^4) = p/R$$

Here p is the pressure difference between two ends of the capillary and R is the fluid resistance.

- Accelerated fluid containers:- $\tan \theta = a_x/g$
- If W be the weight of a body and U be the up thrust force of the liquid on the body then,
- (a) The body sinks in the liquid of W>U
- (b) The body floats just completely immersed if W=U
 - Pressure exerted by a column of liquid of height h:- $P = h\rho g$

Here, ρ is the density of liquid.

• Pressure at a point within the liquid:-

$$P = P_0 + h\rho g$$

Here, P_0 is the atmospheric pressure and h is the depth of point with respect to free surface of liquid.